## Ring-Fluorinated Isoquinoline and Quinoline Synthesis: Intramolecular Cyclization of *o*-Cyano- and *o*-Isocyano- $\beta$ , $\beta$ -difluorostyrenes

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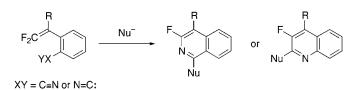
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o-Cyano- $\beta_i\beta$ -difluorostyrenes react with organolithiums selectively at the cyano carbon to generate the corresponding sp<sup>2</sup> nitrogen anions, which in turn undergo intramolecular replacement of the vinylic fluorine to afford 3-fluoroisoquinolines. Similarly, the reaction of  $\beta_i\beta$ -difluoroo-isocyanostyrenes with organomagnesiums or -lithiums generates the corresponding sp<sup>2</sup> carbanions on the isocyano carbon. Subsequent cyclization via substitution of the fluorine leads to 3-fluoroquinolines.

The quinoline and isoquinoline nuclei are regioisomers of benzene-annulated pyridines (benzo[b]pyridine and benzo-[c]pyridine). They are widespread in the alkaloid family and constitute an important class of compounds in pharmaceuticals, agrochemicals, and dyestuffs.<sup>1</sup> Due to their substantial applicability, the syntheses of quinolines and isoquinolines are extensively studied topics.<sup>2,3</sup> However, despite their common properties and much research on their preparation, a synthetic methodology based on a single concept that can be applied to both ring systems remains to be developed.

In our recent publications, we have reported the construction of five- and six-membered ring-fluorinated heterocyclic compounds such as indoles, benzo[*b*]furans, benzo[*b*]fhiophenes, 2-pyrrolines, 2,3-dihydrofurans, 2,3-dihydrothiophenes, isochromenes, and isothiochromenes (Scheme 1a).<sup>4</sup> gem-Difluoroalkenes bearing functional groups such as NHTs, OH, and SH undergo nucleophilic 5- and 6-endo-trig cyclizations via deprotonation of these functional groups that generates sp<sup>3</sup> nitrogen, oxygen, and sulfur nucleophiles. The reactions are promoted by the unique reactivity of gem-difluoroalkenes toward nucleophilic substitution of their

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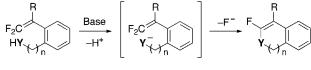
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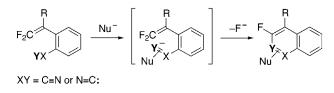
Scheme 1. Intramolecular Substitution of in Situ-Generated Nucleophiles (Y<sup>-</sup>)

(a) sp<sup>3</sup> Nucleophiles Generated by Deprotonation



Y = NTs, O, or S; n = 0 or 1

(b) sp<sup>2</sup> Nucleophiles Generated by Addition to Multiple Bonds



vinylic fluorines via addition—elimination processes.<sup>5</sup> Their reactivity is due to (i) the electrophilic activation of the C–C double bond by the two fluorine atoms, (ii) the stabilization of the intermediary carbanion by the  $\beta$ -anion stabilizing effect of fluorine, and (iii) the leaving-group ability of the fluoride ion. This "intramolecular substitution" concept for ring formation prompted us to explore its application to the construction of quinoline and isoquinoline frameworks.

Herein we wish to report a facile, common methodology for the syntheses of 3-fluorinated isoquinolines and quinolines starting from *ortho*-substituted  $\beta$ , $\beta$ -difluorostyrenes. Although these selectively ring-fluorinated heterocycles have significant potential as components<sup>6a</sup> and synthetic intermediates<sup>6b</sup> of biologically active substances and advanced materials,<sup>7</sup> their synthetic methods are still quite limited.<sup>8,9</sup> The introduction of fluorine atoms especially onto heterocyclic ring carbons is even more difficult than the fluorine introduction onto the carbons of fused benzene rings.

To accomplish direct construction of six-membered aromatic nuclei, we chose to examine  $sp^2$  nucleophiles for the intramolecular substitution of *gem*-difluoroalkenes. It was

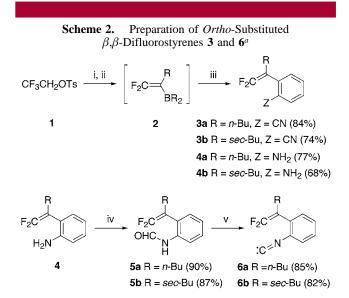
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The starting materials were easily prepared as outlined in Scheme 2 by using the one-pot sequence that we have



<sup>*a*</sup> Reagents and conditions: (i) *n*-BuLi (2.1 equiv), THF, -78 °C, 0.5 h; (ii) BR<sub>3</sub> (1.1 equiv), THF, -78 °C, 1 h, and then rt, 3 h; (iii) *o*-IC<sub>6</sub>H<sub>4</sub>CN or *o*-IC<sub>6</sub>H<sub>4</sub>NHMg*n*-Bu (0.9 equiv), CuI (1.0 equiv), Pd<sub>2</sub>(dba)<sub>3</sub>·CHCl<sub>3</sub> (0.02 equiv), PPh<sub>3</sub> (0.08 equiv), THF–HMPA (4: 1), rt, 1 h; (iv) HCO<sub>2</sub>H (1.2 equiv), Ac<sub>2</sub>O (1.2 equiv), pyridine, rt, 1 h; (v) POCl<sub>3</sub> (1.2 equiv), Et<sub>3</sub>N (2.2 equiv), CH<sub>2</sub>Cl<sub>2</sub>, 0 °C, 0.5 h.

previously developed for the preparation of  $\beta$ , $\beta$ -difluorostyrenes.<sup>12</sup> The coupling reactions of 2,2-difluorovinylboranes **2** [generated in situ from 2,2,2-trifluoroethyl *p*-toluenesulfonate (**1**)] with *o*-iodobenzonitrile or *N*-butylmagnesio*o*-iodoaniline (generated from *o*-iodoaniline and dibutylmagnesium) were carried out in the presence of CuI and a palladium catalyst to give *o*-cyano- $\beta$ , $\beta$ -difluorostyrenes **3** and *o*-amino- $\beta$ , $\beta$ -difluorostyrenes **4**. Aminostyrenes **4** were subjected to formylation of the nitrogen followed by dehydration of the resulting formamides **5**, leading to the desired  $\beta$ , $\beta$ difluoro-*o*-isocyanostyrenes **6**.

We first attempted the generation of intramolecular nucleophiles in difluorostyrenes 3 by the addition of external

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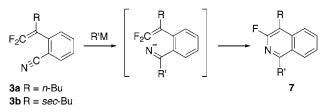
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nucleophiles to the cyano group. In this scheme, the external nucleophile should exclusively attack the cyano carbon without adding to the difluoromethylene carbon. When 3a was treated with *n*-BuMgCl, its nucleophilic attack took place on the cyano carbon to generate the corresponding imino nitrogen anions, which in turn underwent intramolecular substitution of the fluorine to give the expected 3-fluoroiso-quinoline 7a in 65% yield (Table 1, entry 1). While treatment

Table 1.Synthesis of 1,4-Disubstituted 3-Fluoroisoquinolines7 from 3



3	R'M (equiv)	solvent	temp, time	% yield
3a	n-BuMgCl (4.0)	THF	reflux, 10 h	65 ( <b>7a</b> )
3a	<i>n</i> -BuLi/CeCl <sub>3</sub> (3.0)	THF	−78 °C, 3 h	62 ( <b>7a</b> )
3a	<i>n</i> -BuLi (1.2)	THF	−78 °C, 4 h	74 ( <b>7a</b> )
3a	<i>n</i> -BuLi (1.2)	Et <sub>2</sub> O	−78 °C, 0.5 h	86 (7a)
3a	n-BuLi (1.2)	toluene	−78 °C, 1 h	82 ( <b>7a</b> )
3b	n-BuLi (1.2)	Et <sub>2</sub> O	−78 °C, 7 h	82 ( <b>7b</b> )
3a	MeLi (1.2)	toluene	0 °C, 0.5 h	81 ( <b>7c</b> )
3a	t-BuLi (1.2)	Et <sub>2</sub> O	−78 °C, 0.5 h	88 (7 <b>d</b> )
3a	PhLi (1.2)	toluene	−78 °C, 1 h	85 ( <b>7e</b> )
3a	HAl <i>i</i> -Bu <sub>2</sub> (1.1)	toluene	90 °C, 7 h	84 ( <b>7f</b> )
	3a 3a 3a 3a 3a 3b 3a 3a	3a     n-BuMgCl (4.0)       3a     n-BuLi/CeCl <sub>3</sub> (3.0)       3a     n-BuLi (1.2)       3a     n-BuLi (1.2)       3a     n-BuLi (1.2)       3b     n-BuLi (1.2)       3a     MeLi (1.2)       3a     MeLi (1.2)       3a     MeLi (1.2)       3a     PhLi (1.2)       3a     PhLi (1.2)	3a     n-BuMgCl (4.0)     THF       3a     n-BuLi/CeCl <sub>3</sub> (3.0)     THF       3a     n-BuLi (1.2)     THF       3a     n-BuLi (1.2)     THF       3a     n-BuLi (1.2)     Et <sub>2</sub> O       3a     n-BuLi (1.2)     toluene       3b     n-BuLi (1.2)     Et <sub>2</sub> O       3a     MeLi (1.2)     Et <sub>2</sub> O       3a     MeLi (1.2)     Et <sub>2</sub> O       3a     PhLi (1.2)     toluene       3a     t-BuLi (1.2)     toluene	3a $n$ -BuMgCl (4.0)     THF     reflux, 10 h       3a $n$ -BuLi/CeCl <sub>3</sub> (3.0)     THF $-78$ °C, 3 h       3a $n$ -BuLi/CeCl <sub>3</sub> (3.0)     THF $-78$ °C, 3 h       3a $n$ -BuLi (1.2)     THF $-78$ °C, 4 h       3a $n$ -BuLi (1.2)     Et <sub>2</sub> O $-78$ °C, 0.5 h       3a $n$ -BuLi (1.2)     toluene $-78$ °C, 7 h       3a     MeLi (1.2)     Et <sub>2</sub> O $-78$ °C, 0.5 h       3a $n$ -BuLi (1.2)     toluene     0 °C, 0.5 h       3a $t$ -BuLi (1.2)     Et <sub>2</sub> O $-78$ °C, 0.5 h       3a $t$ -BuLi (1.2)     Et <sub>2</sub> O $-78$ °C, 0.5 h       3a $t$ -BuLi (1.2)     Et <sub>2</sub> O $-78$ °C, 0.5 h       3a $t$ -BuLi (1.2)     Et <sub>2</sub> O $-78$ °C, 0.5 h

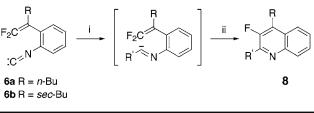
of **3a** with *n*-BuLi/CeCl<sub>3</sub>, an effective reagent for 1,2-additon to  $\alpha,\beta$ -unsaturated ketones,<sup>13</sup> afforded **7a** in 62% yield (entry 2), the reaction of **3a** with *n*-BuLi gave an even better result (74% yield, entry 3). Screening of solvents in this reaction revealed that in Et<sub>2</sub>O or toluene, the addition of *n*-BuLi occurred regioselectively at the cyano carbon of **3a** to afford **7a** in 86 and 82% yields (entries 4 and 5).

Under the reaction conditions obtained above, we examined the reaction of other nucleophiles such as methyllithium, *tert*-butyllithium, phenyllithium, and diisobutylaluminum hydride. As summarized in Table 1, treatment of **3** with these nucleophiles successfully led to cyclization to afford high yields of 3-fluoroisoquinolines **7a**–**f** bearing a butyl, methyl, *tert*-butyl, or phenyl group or hydrogen at the 1-position and a butyl or *sec*-butyl group at the 4-position (entries 4–10). Thus, this sequence provides a facile entry to 1,4-disubstituted 3-fluoroisoquinolines.

After having accomplished the synthesis of isoquinolines as described above, we next tried to construct the quinoline framework by means of the same strategy with the reversed order of the carbon and nitrogen in the *ortho* substituent. For this purpose,  $\beta$ , $\beta$ -difluorostyrenes bearing an isocyano group instead of a cyano group should be employed as starting materials. The reaction of *o*-isocyanostyrenes with nucleophiles was expected to proceed in a similar manner via the corresponding  $sp^2$  carbanion, providing 3-fluoroquinolines with a substituent at the 2-position.

Treatment of isocyanostyrene **6a** with *n*-BuLi in THF gave a complex mixture, probably due to its high nucleophilicity. After the reactions of **6a** with a Grignard reagent were conducted in several solvents, the choice of solvent was found to be crucial in controlling the reaction site. When **6a** was treated with *n*-BuMgBr in THF and toluene, o-(1-butyl-2,2-difluorovinyl)-*N*-pentylideneaniline was obtained in 10 and 73% yields, respectively. These results indicated that the addition of Grignard reagent in toluene selectively occurred on the isocyano carbon, but subsequent cyclization did not take place. To raise the reactivity of the intermediary carbanion, HMPA was added to the reaction mixture after generation of the carbanion, leading to the desired 3-fluoroquinoline **8a** in 69% yield (Table 2, entry 1). Several other

**Table 2.** Synthesis of 2,4-Disubstituted 3-Fluoroquinolines **8** from  $6^a$ 



entry	6	R'M (equiv)	% yield
1	6a	<i>n</i> -BuMgBr (1.2)	69 ( <b>8a</b> )
2	6b	<i>n</i> -BuMgBr (1.2)	60 ( <b>8b</b> )
3	6a	EtMgBr (1.2)	59 ( <b>8c</b> )
4	6a	<i>i</i> -PrMgCl (1.2)	64 ( <b>8d</b> )
$5^b$	6a	t-BuLi (1.2)	78 ( <b>8e</b> )
<b>6</b> <sup>c</sup>	6a	Et <sub>3</sub> GeLi (1.5)	61 ( <b>8f</b> )

<sup>*a*</sup> Reagents and conditions: (i) R'M, toluene, rt, 15 min; (ii) toluene– HMPA (5:1), 0 °C, 1 h, and then rt, 1 h. <sup>*b*</sup> Conditions: -78 °C, 1 h, and then rt, 1 h. HMPA was not added. <sup>*c*</sup> Solvent: toluene–THF (6:1). Conditions: -78 °C, 1 h, and then rt, 4 h.

2,4-disubstituted 3-fluoroquinolines 8b-e were successfully obtained on treatment of 6 with primary, secondary, or tertiary alkylmetals such as *n*-BuMgBr, EtMgBr, *i*-PrMgCl, and *t*-BuLi (entries 2–5).

In the attempted synthesis of 2-silylated 3-fluoroquinolines by the reaction of **6a** with triphenylsilyllithium or dimethylphenylsilyllithium as the external nucleophile, 4-butyl-3fluoroquinoline was obtained instead in 50 or 46% yield, respectively, probably due to the instability of 2-silyl-3fluoroquinolines. We also examined a similar reaction of **6a** with triethylgermyllithium, prepared in situ from triethylgermanium hydride, *N*,*N*,*N'*,*N'*-tetramethylethylenediamine, and *t*-BuLi.<sup>14</sup> The expected addition to the isocyano group and the subsequent cyclization occurred, leading to 3-fluoroquinoline **8f** with a germyl group at the 2-position (entry 6).<sup>15</sup> As shown in Table 2, diverse 2,4-disubstituted 3-fluoroquinolines can be synthesized by this method.

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In conclusion, we have accomplished the efficient construction of ring-fluorinated isoquinoline and quinoline frameworks starting from *ortho*-substituted  $\beta$ , $\beta$ -difluorostyrenes via (i) the addition of nucleophiles to generate intramolecular sp<sup>2</sup> nucleophiles and (ii) their cyclization by substitution of the vinylic fluorine. Both 3-fluoroisoquinolines and 3-fluoroquinolines with a variety of substituents are readily synthesized by this methodology on the basis of the same concept. Thus, we have disclosed a new aspect of fluorine as a synthetic tool for heteroaromatic ring-forming reactions. Acknowledgment. This work was financially supported by a grant from TOSOH F-TECH, INC., to J.I.

Supporting Information Available: Experimental procedures and spectroscopic data of compounds 3-8. This material is available free of charge via the Internet at http://pubs.acs.org.

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<sup>(15)</sup> Reaction of 6a with tributylstannyllithium afforded 4-butyl-3-fluoroquinoline or the corresponding biquinoline, and these results will be reported in detail elsewhere.